

Extended Hubbard model on a C₂₀ molecule

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The electronic correlations on a C₂₀ molecule, as described by an extended Hubbard Hamiltonian with a nearest neighbor Coulomb interaction of strength V , are studied using quantum Monte Carlo and exact diagonalization methods. For electron doped C₂₀, it is known that pair-binding arising from a purely electronic mechanism is absent within the standard Hubbard model ($V = 0$). Here we show that this is also the case for hole doping for $0 < U/t \leq 3$ and that, for both electron and hole doping, the effect of a non-zero V is to work against pair-binding. We also study the magnetic properties of the neutral molecule, and find transitions between spin singlet and triplet ground states for either fixed U or V values. In addition, spin, charge and pairing correlation functions on C₂₀ are computed. The spin-spin and charge-charge correlations are very short-range, although a weak enhancement in the pairing correlation is observed for a distance equal to the molecular diameter.

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I. INTRODUCTION

Shortly after the discovery of superconductivity in C₆₀, it was suggested by Chakravarty, Kivelson and Gelfand^{1,2,3} that an electronic mechanism, in which pairs of electrons preferentially reside on a single molecule rather than on neighboring molecules, might provide the pairing mechanism for superconductivity. Using second order perturbation theory they found evidence for pair binding, above a threshold value of $U/t \approx 3$. They also found that this attraction between doped electrons is accompanied by a violation of Hund's rule, which requires maximal spin, for the two-electron-doped C₆₀, and that for $U/t > 3$, the ground state for two-electron-doped C₆₀ has spin zero.^{1,2} However, recent calculations,⁴ using quantum Monte Carlo (QMC) techniques, suggest that the repulsive Hubbard model does not lead to pairing on C₆₀. On the other hand, there are geometries where pair binding is known to occur^{3,5}. In particular, White *et al.* in exact diagonalization (ED) studies of the extended Hubbard model on the much smaller C₁₂ (truncated tetrahedron) molecule have shown that a negative pair-binding energy (effective attraction between doped electrons) exists for an intermediate value of the on-site Coulomb interaction U [see Eq. (1) and Fig. 3 (a)]. A more realistic model of the fullerenes would include longer ranged Coulomb repulsions, and it was found that this pairing energy also survives in C₁₂ for modestly repulsive values of the nearest-neighbor (NN) interaction, V , but increasing V eventually kills the pair binding. The same violation of Hund's rule as in C₆₀ was also observed in C₁₂ [see Ref. 5 and Fig. 3 (b)].

With a different extended Hubbard model, Sondhi *et al.*⁶ studied the effects of both NN interaction V and the off-diagonal interactions on the pair-binding energy and Hund's rules violation in the C₆₀ molecule. Using perturbative calculations, they find that the NN interaction V terms suppress pair binding while the off-diagonal terms enhance it. Goff and Phillips^{7,8} considered the effects of

both NN interaction V and longer-range terms, V , on the pair-binding energy, again by perturbation theory, and also found that the inclusion of V terms strongly suppresses pair binding in C₆₀.

The fact that ED studies found pair-binding for the smaller C₁₂ molecule⁵ and the recent rapid development of experimental techniques for the synthesis of C₂₀ solid phases^{9,10} make it interesting and timely to explore correlation effects in C₂₀, the smallest gas-phase fullerene molecule which has dodecahedral geometry.¹¹ In Ref. 12, we briefly reported on pair-binding for electron-doped C₂₀ for a wide range of values of $U/t \leq 100$, but with $V = 0$, using both QMC for $U/t \leq 3$ and ED for the full range of values. Using cluster perturbation theory^{13,14} we also identified a metal-insulator transition near $U_c/t \sim 4.2$ for molecular solids formed of C₂₀. In this paper, we provide further details of our numerical techniques and consider both electron and hole doping for an extended Hubbard model with both on-site and NN repulsion. We also study density-density, spin-spin and pairing correlation functions as a function of separation on the molecule.

The extended Hubbard Hamiltonian on a single C₂₀ molecule is defined as

$$H = -t \sum_{\langle ij \rangle \sigma} (c_{i\sigma}^\dagger c_{j\sigma} + h.c.) + U \sum_i n_{i\uparrow} n_{i\downarrow} + V \sum_{\langle ij \rangle} n_i n_j, \quad (1)$$

where $c_{i\sigma}^\dagger$ ($c_{i\sigma}$) is an electron creation (annihilation) operator on site i , indices i, j run over 20 sites of a dodecahedron, U is the on-site Coulomb interaction, V is the NN Coulomb interaction, and $n_i = n_{i\uparrow} + n_{i\downarrow}$ is the number of electrons on site i . Our goal is here to focus on strong correlation effects in C₂₀ using exact numerical techniques. The Hamiltonian Eq. (1) is a simplified model of C₂₀ but it still largely captures such correlation effects. We calculate ground state energies as a function of both U and V for neutral, one- and two-electron dopings. Comparisons among these energies show that the electronic

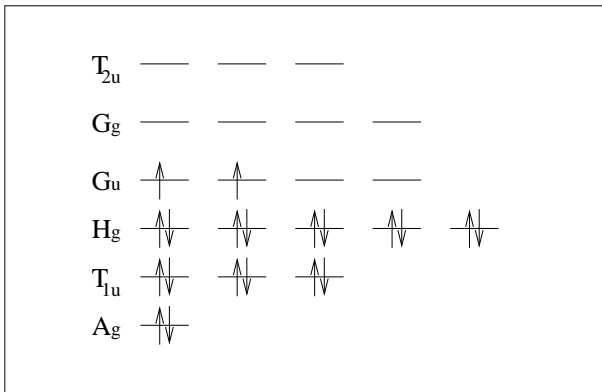


FIG. 1: Huckel molecular orbitals of a neutral dodecahedral C_{20} molecule.¹⁵

pair-binding energy $\Delta_b(21) = E(20) + E(22) - 2E(21)$ is positive (repulsive) for the parameter ranges studied ($0 < U/t \leq 3$ for $V/t = 0.2$ and $0.20 \leq V/t \leq 0.46$ for $U/t = 1$). This implies that it is energetically favorable for two electrons to stay on different molecules as opposed to the same molecule. We also find that the existence of a NN Coulomb interaction V enhances this tendency, as expected, in order to reduce the intramolecular Coulomb interaction energy. For hole doping, the corresponding hole pair-binding energy $\Delta_b(19) = E(18) + E(20) - 2E(19)$ is again positive (repulsive) for the parameter range ($0 < U/t \leq 3$ and $V = 0$), i.e., there is an effective repulsion between two doped holes on the same C_{20} molecule.

Unlike the case of C_{60} , the highest occupied molecular orbital (HOMO) of the neutral C_{20} molecules, in the weakly interacting limit, is a four-fold orbitally degenerate level occupied by two electrons. (See Fig. 1.) Hund's rules predict for this case that the two electrons occupy different orbitals and have total $S = 1$, implying that, in the absence of a Jahn-Teller distortion, the neutral molecule has a magnetic moment. In previous work¹², for $V = 0$, we have confirmed this magnetic moment for $0 < U/t < 3$ and shown that at the metal-insulator transition, U_c , the ground-state changes from a spin triplet to a singlet for neutral C_{20} and from $S = 2$, through $S = 1$, to $S = 0$ for C_{20}^{2-} . Here we extend this analysis to determine ground state spin configuration for neutral C_{20} for a fixed value of $U/t = 2$ as a function of V , and find a level crossing between $V/t = 1$ and $V/t = 1.5$ for spin triplet and singlet states. For $U/t = 2$, we estimate the critical V_c/t to be 1.1 for the spin triplet to singlet transition of the neutral molecule. In light of our results for $V = 0$, we expect that, in this case too, the magnetic transition at V_c will coincide with a metal-insulator transition for molecular solids formed of C_{20} . We also investigate the pair-binding energy for the hole doped case for both $V = 0$ and $V \neq 0$, and examine the effect of a non-zero V on Hund's rule.

The occurrence of orbital degeneracy and the resulting magnetic moment are tied to the icosahedral symmetry

of the molecule. Simple molecular orbital calculations strongly suggest that the molecular symmetry is lowered by a Jahn-Teller effect from I_h to D_{3d} , with the HOMO being a non-degenerate singlet.¹⁶ However, the correlation effects that give rise to Hund's rule compete with this tendency to form a singlet ground state, and hence they also compete with the Jahn-Teller effect. As reported previously,¹² we find that when the on-site Coulomb interaction U/t is sufficiently large ($U \gtrsim 4.2t$), the ground state is gapped with $S = 0$ and the I_h symmetry is likely stable against a D_{3d} distortion. In order to more exclusively focus on the effects of the non-zero V term we shall here assume that the icosahedral symmetry is unbroken even for smaller U values.

In the next section, we briefly introduce the projection quantum Monte Carlo (PQMC)¹⁷ and ED methods for this model. This is followed, in Section III, by a comparison of PQMC with ED results on a C_{12} and a discussion of Hund's rule violation in C_{12} . In section IV we focus on the C_{20} molecule. Hole pair-binding in C_{20} is discussed and the influence of a non-zero next nearest neighbor V on the pair-binding is investigated and results for the triplet-singlet transition with V are described along with calculations of several correlation functions in the C_{20} molecule. Section V contains discussion and conclusions.

II. METHOD

A. PQMC

As noted in Ref. 18, the idea in PQMC simulations of the extended Hubbard model is to decouple the two-body interaction terms (both U and V terms) in the partition function by means of discrete Hubbard-Stratonovich transformations.¹⁹ The resultant one-body terms are coupled to several auxiliary Ising spin fields that live either on the lattice sites (U term) or on the lattice bonds (V term). One such discrete transformation in the V term is given by

$$e^{-\Delta\tau V n_{i\alpha} n_{j\beta}} = \frac{1}{2} \text{Tr}_{\{\sigma_{ij}^{\alpha\beta}\}} e^{\lambda_2 \sigma_{ij}^{\alpha\beta} (n_{i\alpha} - n_{j\beta}) - \frac{\Delta\tau V}{2} (n_{i\alpha} + n_{j\beta})}, \quad (2)$$

where $\alpha, \beta = \uparrow, \downarrow$, $\sigma_{ij}^{\alpha\beta} = \pm 1$ is the auxiliary Ising spin on bond (i, j) , $\Delta\tau$ is the discrete imaginary time slice in PQMC, and the parameter λ_2 is determined by $\tanh^2(\lambda_2/2) = \tanh(\frac{\Delta\tau V}{4})$. The same decoupling equation applies for on-site Coulomb interactions, i.e., $i = j$, except that the constant V is replaced by U and λ_2 by λ_1 , which is similarly given by $\tanh^2(\lambda_1/2) = \tanh(\frac{\Delta\tau U}{4})$. These one-body fermionic terms in the partition function can then be explicitly traced out, leaving traces over the auxiliary Ising spins, which can be evaluated by Monte

Carlo (MC)¹⁹.

$$\begin{aligned}
 Z &= \sum_{\{\sigma\}} \prod_{\alpha} \det[1 + B_L(\alpha) B_{L-1}(\alpha) \cdots B_1(\alpha)] \\
 &= \sum_{\{\sigma\}} \det O(\{\sigma\})_{\uparrow} \det O(\{\sigma\})_{\downarrow},
 \end{aligned} \quad (3)$$

where $\{\sigma\} = \{\sigma^1, \sigma^2, \sigma^3, \sigma^4, \sigma^5\}$ is the set of five species of Ising fields, with σ^1 representing the on-site Ising spins and σ^{2-5} the NN bond Ising spins (one for each of the 4 spin configurations). The B_l matrices are defined as

$$B_l(\alpha) = e^{-\Delta\tau K/2} e^{W^\alpha(l)} e^{-\Delta\tau K/2}, \quad (4)$$

$$(K)_{ij} = \begin{cases} -t & \text{for } i, j \text{ NN,} \\ 0 & \text{otherwise,} \end{cases} \quad (5)$$

$$W_{ij}^\alpha(l) = \alpha[\delta_{ij} \lambda_1 \sigma_i^1(l) + \delta_{\langle ij \rangle} \lambda_2 \sum_{m=2}^5 \sigma_{ij}^m], \quad (6)$$

$$\delta_{\langle ij \rangle} = \begin{cases} 1 & \text{for } i, j \text{ NN,} \\ 0 & \text{otherwise,} \end{cases} \quad (7)$$

where $l = 1, \dots, L$ is the time slice index, and $\alpha = \pm 1$ denotes the two determinants in Eq. (3).

A complete MC sweep through the lattice will therefore consist of trial flipping of one species of auxiliary Ising spins on all the lattice sites and trial flipping of four species of auxiliary Ising spins on all the NN bonds in the lattice system. Fast calculation of the probability ratio in flipping one bond Ising spin at one time slice is still possible using the local update technique,²⁰ except that one needs to apply the probability ratio formula twice for each bond Ising spin flip (which affects two sites).

We remark that in this decomposition scheme it is possible to treat even longer range Coulomb interactions [e.g., next nearest neighbor (NNN) Coulomb interactions, etc.] by introducing more species of auxiliary Ising spins that live on these longer bonds. The only problem is that one needs to walk through a larger and larger phase space of the auxiliary Ising spins during the MC simulations, which will, of course, increase the computation time. Practically, we find that, to collect the same amount of data, the CPU time doubles for $V \neq 0$ compared with the $V = 0$ case.

In a typical calculation the projection factor β in PQMC was taken to be $\beta = 10/t$, and the discrete time slice was set at $\Delta\tau = 0.05/t$. 10^3 MC warm-up sweeps through the whole space-time lattice are typically performed before collecting data. To estimate the statistical errors, we use the same method as was used in Ref. 4.

B. Exact Diagonalizations (ED)

The exact diagonalizations on C_{12} are done using standard Lanczos techniques and we therefore focus on the ED of C_{20} . We always use total particle number N and total S_z as quantum numbers since they are conserved

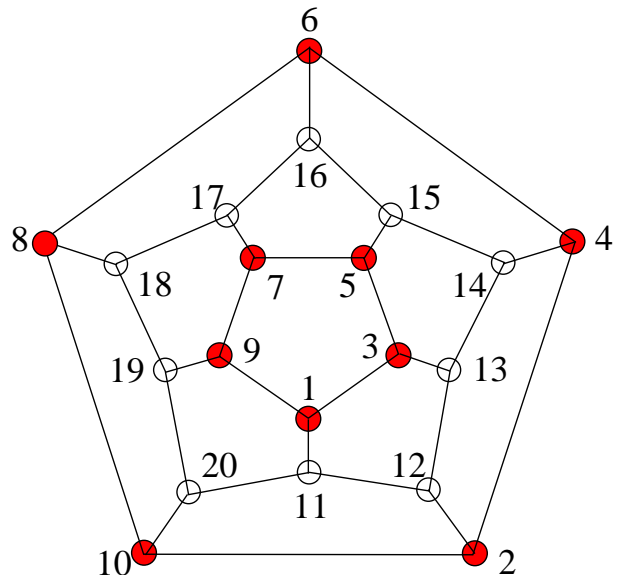


FIG. 2: (Color online) Dodecahedral C_{20} geometry in 2D view. Solid and empty points denote two sets (orbits) of carbon atoms divided by the S_{10} symmetry.

and we perform ED in the corresponding reduced Hilbert space. In addition, ED are performed using the S_{10} subgroup symmetry present in the point group I_h . The improper rotations generated by the elements of S_{10} can be visualized as a rotation of an angle $2\pi/10$ around the center of a pentagon followed by a reflection in a plane perpendicular to the rotation axis. This is illustrated in Fig. 2 where the numbering of the sites is to be understood in the following way: the sites 1 through 10 are shifted up by 1 (modulo 10) under S_{10} and the sites 11 through 20 are shifted in a similar manner. Hence, under the action of the S_{10} group two different orbits exist, marked by the solid and open points in Fig. 2. Many other symmetries exist but the S_{10} symmetry is large and relatively easy to implement, and we have not exploited additional symmetries since the added cpu-time needed to implement them was significant enough to offset the time gained from reducing the size of the Hilbert space. The S_{10} quantum number can be thought of as a pseudo angular momentum, j_{10} , and for each value of N and S_z we have to find the value of j_{10} that corresponds to the ground-state. In many cases it is not an obvious value and it is often non-zero. In the accompanying tables we show the values of j_{10} corresponding to the listed energies and in Table IV we show complete dispersion of the lowest magnetic modes for neutral C_{20} as a function of j_{10} .

The calculations are fully parallelized Lanczos calculations executed on SHARCNET computers. A typical calculation performed at half-filling for $N = 20$, $S_z = 0$ that, after S_{10} symmetry reductions, requires a Hilbert space of $\mathcal{N} = 3,418,725,024$, is performed with $P = 64$ cpu's using about 540 seconds of cpu-time (for each cpu)

per Lanczos iteration. The memory requirement for this example is roughly 2.1Gb per cpu. Excellent convergence is always observed with less than 300 Lanczos iterations, typically less than 200.

The heart of the Lanczos calculation is the matrix vector multiplication that in this case has to be implemented in parallel. As one of several choices, we have chosen to have each cpu apply the full matrix to one section of the vector with each cpu returning the corresponding section of the resulting vector. The partial results from each cpu therefore needs to be communicated between all P processors with each processor communicating to all others. Due to the size of the involved Lanczos vectors (40-60Gb) which greatly exceeds the available per-cpu memory, it is necessary to repeat this $P \times P$ communication step many thousands of times per Lanczos step. The communication step therefore quickly becomes the bottle-neck in the calculation unless it can be done very efficiently. Fortunately, this is possible using non-blocking communications where the individual cpu's do not wait for a communication to complete. The drawback of using non-blocking communications is that buffer space has to be allocated until it has explicitly been verified that the communication has been completed. We have implemented a dual buffer strategy yielding an extremely efficient communication step. The cpu-time spent per cpu is, for all accessible number of processors we have been able to check, overwhelmingly dominated by actual calculations rather than communications. For a fixed \mathcal{N} we have then observed almost linear scaling for $P = 64, 128, 256, 384$ and 512. The great advantage of this approach is that the complexity of Lanczos calculations scale with the size of the Hilbert space, \mathcal{N} , as $\mathcal{N} \log \mathcal{N}$. Neglecting the logarithm, a doubling of the size of the Hilbert space, \mathcal{N} , can then be almost compensated by doubling the number of processors P .

	S	S_z	ED	PQMC	sign
E_{12}	0	0	-9.4647669965	-9.466(2)	0.97
E_{13}	1/2	1/2	-6.8287003500	-6.829(4)	0.33
E_{13}	3/2	3/2	-6.0844214907	-6.059(6)	0.20
E_{14}	0	0	-4.1568425864	-4.11(1)	0.11
E_{14}	1	1	-4.0772924523	-4.080(5)	0.34
$\Delta_{1,0}$	(1/2,0)		2.6360666465	2.637(4)	
$\Delta_{1,0}$	(3/2,0)		3.3803455058	3.407(6)	
$\Delta_b(13)$	(0,0,1/2)		0.0357911171	0.08(1)	

TABLE I: Comparison of ED and PQMC calculations on the truncated tetrahedron (12 sites) at $U = 2t$ and $V = 0.2t$. $E_n(S_z)$ is the energy of a system with n electrons and z -component of total spin S_z . $\Delta_{n,m}$ is the energy difference $E_{12+n}(S_z^n) - E_{12+m}(S_z^m)$ with (S_z^n, S_z^m) given in the second column. For binding energies $\Delta_b(n)$ the second column shows $(S_z^{n+1}, S_z^{n-1}, S_z^n)$ - the S_z values for 3 states involved in its calculation⁴

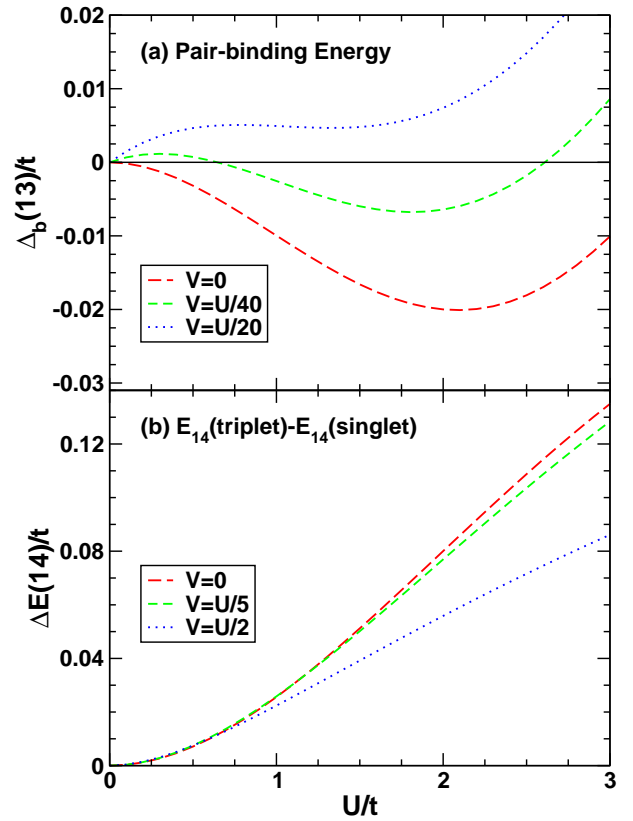


FIG. 3: (Color online) (a) Variation of the pair-binding energy $\Delta_b(13) = E(12) + E(14) - 2E(13)$ of a truncated tetrahedron molecule (C_{12}) with U and V as in Fig. 3 in Ref. 5. (b) Hund's rules violation in the two-electron doped C_{12} molecule, where $\Delta E(14) = E_{14}(\text{triplet}) - E_{14}(\text{singlet})$.

III. RESULTS FOR C_{12}

Before turning our attention to the C_{20} molecule we investigate the simpler C_{12} molecule in the truncated tetrahedron configuration. As mentioned above, previous studies⁵ have found a negative pair-binding energy on this molecule that, however, became positive (repulsive interaction) in the presence of a sufficiently large V . The purpose of this investigation is two-fold. First of all, we want to verify the correctness of our numerical approach while at the same time highlighting some of the subtleties of interpreting the PQMC data. Secondly, due to the relative ease with which calculations can be performed on this molecule it allows for a rather detailed study of the correlation between the negative pair-binding energy and a violation of Hund's rule for the two electron doped molecule^{1,2}.

A. Tests on the C_{12} molecule

To test our ED program, we use the same parameters as in Ref. 5 and we are able to reproduce the same pair-binding energy as shown in Fig. 3 (a). In Table I, we see

good agreement between PQMC and ED energy values within statistical error bounds. An exception is found for E_{14} and $S_z = 0$, where the PQMC result lies a bit higher than the ED energy value. This is due to the mixture of singlet and triplet components in the $S_z = 0$ sector and the near degeneracy of these two states that makes the projection of the singlet ground state out of the mixed state difficult.⁴ We will see that this difficulty *does not* occur for C_{20} , where the ground state with two-electron doping is in the spin-2 sector for $U/t \leq 3$. Hence the pair-binding energy extracted for C_{20} by PQMC for $U/t \leq 3$ is more reliable than the one for C_{12} .

B. Hund's rule violation for C_{12}^{2-}

In the perturbation theory studies of pair-binding in the larger fullerene C_{60} ,^{1,2} it was noted that a negative pair-binding energy (effective attraction) was correlated with a violation of Hund's rule for the two-electron doped molecule; i.e., that for C_{60}^{2-} , the ground state was found to be a singlet. Although our QMC results did not support the existence of pair-binding in C_{60}^{2-} and found a spin-triplet ground state, it is of interest to examine the correlation between pair-binding and the violation of Hund's rule in C_{12} . The non-interacting $V = U = 0$ neutral molecule has completely filled levels and hence a total spin zero. Added electrons therefore enter an unfilled level with an orbital degeneracy of 3. Hund's rule would then predict C_{12}^{2-} to have total $S = 1$. What we find is that the ground state of C_{12}^{2-} is a singlet both when the pairing is attractive and when it is driven repulsive by increasing the nearest neighbor repulsion V . This is shown in Fig. 3(b) where the singlet state is found to lie below the triplet state for both positive and negative pair-binding energies, for the range of U and V studied. Thus, for this case, Hund's rule is found to be violated where pair-binding occurs as well as where it does not.

IV. RESULTS FOR C_{20} MOLECULE

We now turn to the more interesting case of the C_{20} molecule. Compared with the $V = 0$ case, where PQMC already has a sign problem for the non-bipartite dodecahedral molecular geometry, the NN Coulomb interaction V terms introduce more sources of negative probability weight, lowering the average value of the sign. Fig. 4 shows the average sign for these two cases. For the worst case ($N = 20, S_z = 1, U/t = 3, V/t = 0.2$), where the average sign is as low as 0.05, we have collected 7.2×10^7 MC lattice sweeps. This gives a relatively large but nevertheless meaningful error bar. [See Fig. 5 (a).] For other parameter values, we have collected about 2.2×10^7 MC sweeps. The acceptance ratio for the on-site Ising spin trial flipping ranges from 80% ($U/t = 3$) to 93% ($U/t = 1$), while that for the bond Ising spins is about 95% due to the small value of $V/t = 0.2$.

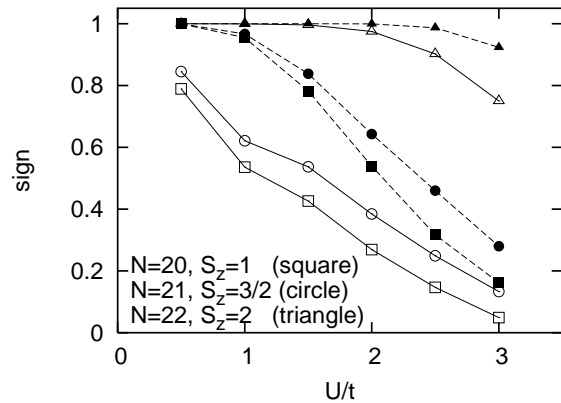


FIG. 4: Average sign behavior for both $V/t = 0$ (solid symbols) and $V/t = 0.2$ (hollow symbols) at different fillings $N = 20, 21, 22$. The lines connecting the points are guides to the eye only.

A. Pair-binding energy

Table II shows the energies of the C_{20} molecule at different fillings from PQMC and ED for $U/t = 2$ and $V = 0$. Both ED and PQMC predict the ground states to be in the same spin sectors for the molecule, and the calculated energies are in agreement within MC error bounds.

In order to understand the comparison of PQMC and ED data in Table II, it is important to recognize a systematic weakness of PQMC which is that, when the ground state is a spin multiplet, the different partners appear to have different energies, increasing with decreasing values of $|S_z|$, because the states with smaller values of $|S_z|$ mix with higher lying states that have the same value of $|S_z|$. In general except for statistical error, a state with $S_z = 0$ will appear to lie above its partners with the same

	S	S_z	ED	j_{10}	PQMC	sign
E_{18}	0	0	-22.4044466933	0	-22.402(1)	1.00
E_{18}	1	1	-21.6778357505	$\pm 3, 5$	-21.637(2)	0.49
E_{19}	1/2	1/2	-21.5223243600	$\pm 1, \pm 3$	-21.5227(6)	0.64
E_{19}	3/2	3/2	-20.8990191757	$0, \pm 4$	-20.826(3)	0.35
E_{20}	1	0	-20.5983834340	$0, \pm 2$	-20.533(3)	0.26
E_{20}	1	1	-20.5983834340	$0, \pm 2$	-20.597(2)	0.54
E_{20}	0	0	-20.5920234654	$0, \pm 2, \pm 4$		
E_{20}	2	2	-19.9634427212	$\pm 2, \pm 4, 5$		
E_{21}	3/2	1/2	-19.6331786587	$\pm 1, \pm 3$	-19.465(8)	0.19
E_{21}	3/2	3/2	-19.6331786587	$\pm 1, \pm 3$	-19.634(1)	0.64
E_{22}	2	0	-18.6289129089	0	-18.282(7)	0.10
E_{22}	2	1	-18.6289129089	0	-18.448(5)	0.32
E_{22}	2	2	-18.6289129089	0	-18.628(1)	1.00

TABLE II: Comparison of ground state energies from ED and PQMC calculations on the C_{20} molecule at $U/t = 2$ and $V = 0$. See the caption in Table I for the corresponding definition of various quantities.

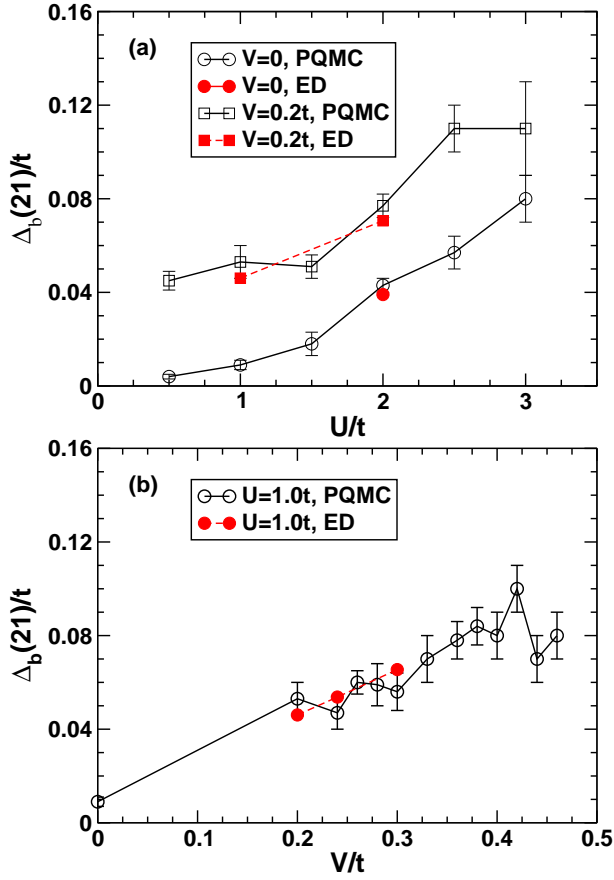


FIG. 5: (Color online) Electronic pair-binding energies $\Delta_b(21)/t$ as a function of U/t and V/t from ED and PQMC simulations. (a) The variation of pair-binding energy with U/t for fixed V/t values. (b) The variation of pair-binding energy with V/t for fixed $U/t = 1$. The lines connecting MC and ED points are guides to the eye only.

total S . This tendency is apparent in the results for E_{20} ($S=1$), E_{21} ($S=3/2$), and E_{22} ($S=2$). Conversely, if a ground state with $S_z = 0$ lies below a state with $S_z = 1$, we expect the ground state to be a singlet. However, in this case, the value of the ground state energy will be perturbed upward by any admixture of the next higher state with $S = 1, S_z = 0$, as happened for $E_{14}(S=0)$ in Table I. In general it is also true that accurate PQMC results are more easily obtained when the average sign is close to 1 compared to when the average sign is small.

Pair-binding energies $\Delta_b(21)/t$ (electron) and $\Delta_b(19)/t$ (hole) as a function of both U/t and V/t are shown in Fig. 5 and 6, respectively. For $V = 0$, for both electron and hole doping, we see that the pair-binding energy is always positive (repulsive) for $U/t > 0$, and increases with increasing U/t . This is the same behavior as we observed for the C_{60} molecule.⁴ Turning on the NN Coulomb interaction V ($V/t = 0.2$ in Fig. 5(a)) increases the pair-binding energy further. Hence, putting two extra electrons on the same neutral molecule becomes more costly when the NN Coulomb interaction is not

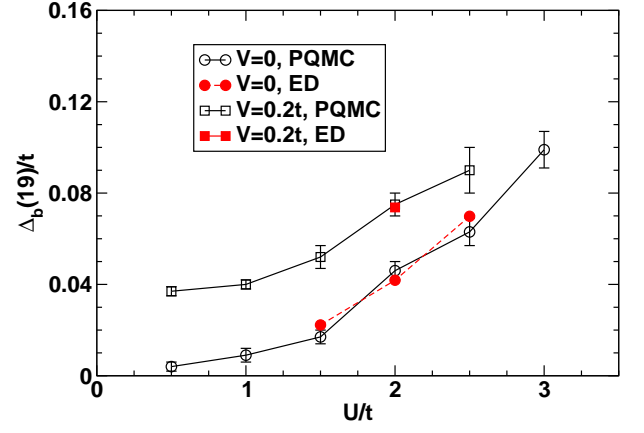


FIG. 6: (Color online) Hole pair-binding energies $\Delta_b(19)/t$ as a function of U/t for $V = 0$ from ED and PQMC simulations. The lines connecting MC points are guides to the eye only.

negligible. Panel (b) in Fig. 5 shows the variation of the pair-binding energy as a function of V/t for fixed $U/t = 1$. Again the pair-binding energy is positive (repulsive), and generally increases with V/t . The agreement between ED and PQMC results is fairly good and even though the PQMC data show some tendency to non-monotonic behavior for this interval of V/t , the ED results show that this is explained by the natural statistical spread of the data. Hence, in the regime $V < V_c$, $U < U_c$, the pair-binding energy increases with both U and V and energetically, it becomes increasingly favorable for two electrons to stay on two different C_{20} molecules. However, we note that, for $V = 0$, $U > U_c$ it was previously found¹² that the pair-binding energy *decreases* with U , reaching a minimum at $U/t \sim 10$, before increasing and reaching a finite value in the $U \rightarrow \infty$ limit.

	S	S_z	$U/t = 3$	S	S_z	$U/t = 5$ (ED)	j_{10}
E_{20}	1	0	-17.04(2)	0	0	-12.111284292	5
E_{20}	1	1	-17.036(6)	1	1	-11.877033283	0, ± 2
E_{21}	3/2	1/2	-15.29(6)	1/2	1/2	-9.1165560273	$\pm 1, \pm 3, 5$
E_{21}	3/2	3/2	-15.529(5)	3/2	3/2	-8.9633623599	$\pm 1, \pm 3$
E_{22}	2	0	-13.936353	1	0	-5.9715313615	$\pm 2, \pm 4$
E_{22}	2	1	-13.81(2)	1	1	-5.9715313615	$\pm 2, \pm 4$
E_{22}	2	2	-13.935(1)				

TABLE III: Ground state energies for neutral, one- and two-electron-doped C_{20} molecules at $U/t = 3, 5$ and $V = 0$ from PQMC and ED, which shows a transition between Hund's and anti-Hund's states at $3 < U/t < 5$ for neutral, one-, and two-electron-doped molecules, respectively. Data without error bars are from ED.

U	S	$j_{10} = 0$	$j_{10} = \pm 1$	$j_{10} = \pm 2$	$j_{10} = \pm 3$	$j_{10} = \pm 4$	$j_{10} = 5$
2	0	-20.5920234655	> -19.90	-20.5920234655	> -19.90	-20.5920234655	-20.0527029539
	1	-20.5983834340	-19.9776970001	-20.5983834340	-19.9776970001	-20.5981592741	-19.9634427213
5	0	-12.0123014488	-11.6726562451	-12.0123014488	-11.6726562451	-12.0123014488	-12.1112842959
	1	-11.8770332831	-11.8472120431	-11.8770332831	-11.8472120431	-11.8103044760	-11.8118179567
8	0	-8.0452584717	-7.806831	-8.0452584717	-7.8068365859	-8.0452584717	-8.1803385740
	1	-7.9497836200	-7.9415479844	-7.9497836200	-7.9415479844	-7.8490047592	-7.9156714009

TABLE IV: ED results for the dispersion of the lowest singlet and triplet states with j_{10} for neutral C_{20} with $U/t = 2, 5$ and 8 in all cases with $V/t = 0$.

B. Hund's rule

It is also clear, from the data in Table II and III, that Hund's rule is obeyed for the corresponding range of parameters, i.e., $U/t \leq 3$, $V = 0$. That is, the ground state for 20 through 22 all have the maximum values of total spin for electrons outside the C_{20}^{2+} core, ranging from total spin 1 for 20 electrons through total spin 2 for 22 electrons. This behavior occurs in the range of parameters where PQMC converges (for maximal $|S_z|$ as discussed above.) As U/t is increased above 3, the sign problem prevents reliable PQMC calculations. This difficulty does not arise in ED where accurate calculations are possible for essentially any value of U/t . We have used ED to explore what happens for larger values of U/t .¹² For example, results for $U = 5t$ are shown in the right hand columns of Table III. Here Hund's rule is clearly violated. For 20 electrons, the ground state has spin zero; for 21 electrons the ground state has spin 1/2; while for 22 electrons the ground state has spin 1. Clearly there are level crossings in the range $3 < U/t < 5$. Additional results in this regime are given in Ref. 12. ED also allows the calculation of the spin gap, the gap between the ground state and the lowest lying excited state with different total spin. Results are shown in Table IV for a neutral C_{20} molecule with $U = 2, 5, 8$ and $V = 0$. When the metal-insulator transition occurs in the vicinity of $U_c/t \sim 4.2$, the ground-state spin changes from an orbitally degenerate $S = 1$ for $U < U_c$ to a non-degenerate singlet for $U > U_c$. From the results presented in table IV we see that it is the singlet state at $j_{10} = 5$ that moves toward the bottom of the spectrum with increasing U and eventually, for $U > U_c$ becomes the ground-state. Focusing on the case $V/t = 0$, we see from table IV that at $U/t = 2$, the ground-state energy is a singlet $E^1/t = -20.5983834340$ with a gap to the lowest lying singlet of $\Delta E^{1,0}/t = 0.0063599685$. Here the superscripts denote the spin of the ground- and excited states, respectively. For $U/t \geq 5$ we find that the ground-state for the I_h configuration now is a non-degenerate singlet, $S = 0$, with energy $E^0/t = -12.1112842922$. The lowest lying triplet excitation with $\Delta E^{0,1}/t = 0.2342510092$. This picture continues to hold for larger U/t with the triplet gap at $U/t = 8$ only slightly larger, $\Delta E^{0,1}/t = 0.2305549540$.

Next we explore the ground state spin of the neutral molecule with different V/t values for a fixed $U/t = 2$.

Using ED techniques we determine that the ground-state for $V/t = 1$ and $V/t = 1.5$ in both cases occur for $j_{10} = 0$. However, the ground state changes from a spin triplet for $V/t = 1$ to a spin singlet for $V/t = 1.5$. Specifically, we find at $V/t = 1$, $E(\text{Singlet})=5.702018$ and $E(\text{triplet})=5.639496$, whereas for $V/t = 1.5$ we find $E(\text{Singlet})=17.318536$ and $E(\text{triplet})=17.499741$. By assuming a linear dependence of the energy on V/t in this region, we determine that the level crossing occurs near $V_c/t \sim 1.1$ for $U/t = 2$.

C. Correlation functions

We have also investigated what other correlations might be induced in the C_{20} molecule by calculating the following correlation functions: charge-charge, spin-spin, and pairing correlations as a function of lattice distance. Similar calculations for the C_{60} molecule have been reported in Refs. 21 and 22.

We define the correlation functions with respect to lattice site 1 in the neutral molecule: $\langle n_1 n_i \rangle$ is the charge-charge correlation, $\langle S_1 \cdot S_i \rangle$ is the spin-spin correlation, and $\langle c_{1\sigma}^\dagger c_{i,-\sigma}^\dagger c_{i,-\sigma} c_{1,\sigma} \rangle$ is the pairing correlation, where $i = 1, \dots, 20$. In Fig. 7, we show the variation of these correlations for $U = 2t$ and $3t$ as a function of lattice spacing d_{1i}/R , where d_{1i} is the distance between site 1 and i , and R is the molecular diameter. For the dodecahedral geometry, there are only 5 inequivalent neighbors, all at distinct distances. One can understand the on-site correlations in terms of the probabilities, p_n , $n=0,1,2$, for having n electrons on each site. Then, the on-site correlations functions are $\langle n_1^2 \rangle = p_1 + 4p_2$, $\langle S_1^2 \rangle = 3p_1/4$, and $\langle c_{1\sigma}^\dagger c_{1,-\sigma}^\dagger c_{1,-\sigma} c_{1,\sigma} \rangle = p_2$.

In Fig. 7(a) we show results for the charge-charge correlation function for 2 different values of U with $V = 0$. As expected, the on-site charge-charge correlation is reduced by an increase of the on-site Coulomb interaction U [panel (a)]. At larger distances, the charge on site 1 and i are uncorrelated. The unit value of the charge-charge correlation corresponds to uniform distribution of charge.

Fig. 7(b) shows the spin-spin correlation function again for $U/t = 2, 3$ with $V = 0$. The NN spin-spin correlation has a negative finite value, and its magnitude is enhanced by a larger U value. For spatial distances larger than 1

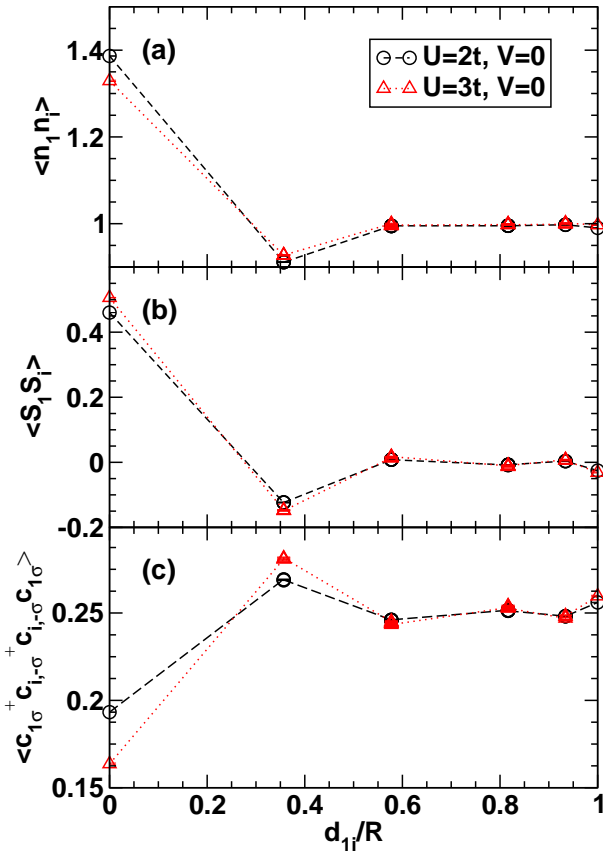


FIG. 7: (Color online) Variation of (a) charge-charge, (b) spin-spin, and (c) pairing correlation functions for $U = 2t, 3t$ and $V = 0$ for a C_{20} molecule with respect to the lattice site spacing. d_{1i} is distance between site 1 and i . R is diameter of C_{20} molecule.

we see that this correlation function quickly approaches 0. Similar behavior has been observed for the spin-spin correlation in the C_{60} molecule in Ref. 21 and it was suggested that the rapid decay of the spin-spin correlation function was indicative of a resonant valence bond (RVB) or “spin dimer” state. The similarity between our results and those of Ref. 21, suggest that the spin correlations in the ground-state of C_{20} also might be described by considering valence bond states including only dimers of relatively short length.

QMC results for the pair correlation are shown in Fig. 7(c) with $U/t = 2, 3$ and $V = 0$. Interestingly, there is a peak of the pairing order when site 1 and i are NN sites. This again supports the RVB or “spin dimer” model for the ground-state. Beyond the nearest-neighbor distance, the pairing correlation function, along with the other correlation functions, is very close to its uncorrelated value except for $d_{1i} = R$, where the pairing order parameter is slightly enhanced. At the same time the spin-spin correlation is slightly negative showing an anti-ferromagnet correlation. This corresponds to the “dumb bell” model proposed in Ref. 22 where electron pairs are

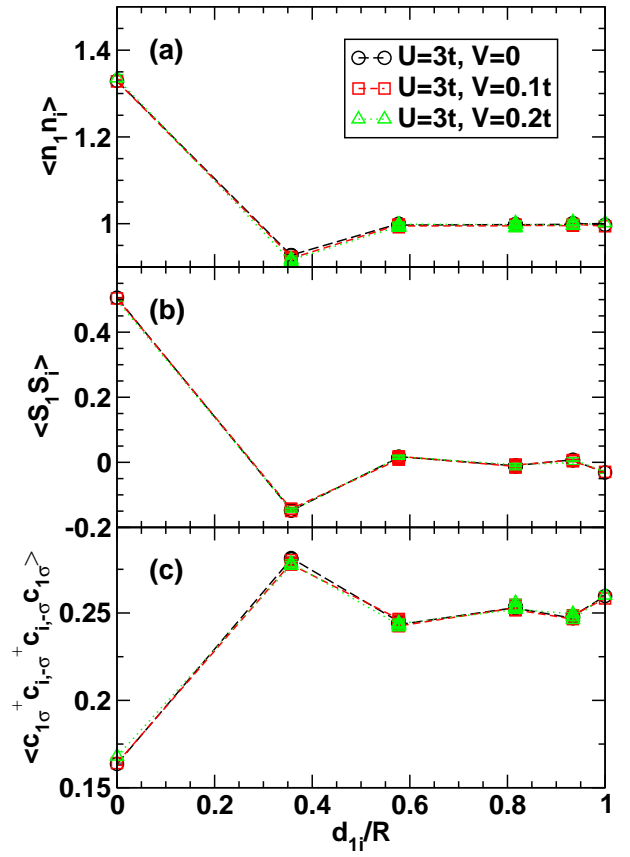


FIG. 8: (Color online) Variation of (a) charge-charge, (b) spin-spin, and (c) pairing correlation functions for $U = 3t$ and $V = 0, 0.1t, 0.2t$ for a C_{20} molecule with respect to the lattice site spacing.

formed at the maximal distances of the molecular diameter. We note that in the present case, the enhancement of the correlations at the distances of R corresponding to this “dumb bell” pairing is relatively weak.

We have also studied the influence of a non-zero V on the correlations. In Fig. 8 we show results for a fixed $U/t = 3$ and three different values of $V = 0, 0.1t, 0.2t$. Clearly, the effect of the NN Coulomb interaction V on these correlation functions is relatively weak, with the curves being almost identical for the range of V considered here.

V. CONCLUSIONS

In this paper we have studied the extended Hubbard model on a C_{20} molecule through ED and PQMC simulations. The comparison clearly elucidates the relative strengths of the two methods. PQMC is possible for much larger systems than can be treated by ED. However, ED has been applied successfully to the Hubbard model on 20 sites with 18-22 electrons, by making effective use of the capabilities of a large number of coupled

processors. PQMC works best when the ground state is well separated from excited states with the same value of S_z . As a result, ground states with larger total spin S and maximal $|S_z|$ are most accurately determined, while ground states with $S = 0$ are sometimes problematic. This behavior was also found in our earlier work on C_{60} ,⁴ and the comparison of ED and PQMC results for C_{20} is consistent with and lends confidence to those earlier results.

The pair-binding energy for C_{20} shows that extra added electrons (holes) prefer to sit on different molecules, rather than to reside in pairs on molecules. This rules out the possibility that the extended Hubbard model on a single C_{20} molecule can produce an effective attraction between electrons (holes) from purely electronic interactions. Our earlier work showed that this conclusion applies to the C_{60} molecule as well.⁴ We also find that Hund's rule is obeyed for $U/t \leq 3$ and small values of V and that larger values of U and V lead to level crossings and ground states for which Hund's rule is violated. For fixed $V = 0$, we have determined that this transition happens between $U/t = 3$ and $U/t = 5$, at $U_c/t \sim 4.2$. And for fixed $U/t = 2$, as a function of V , we have determined that this transition happens between $V/t = 1$ and $V/t = 1.5$, at $V_c \sim 1.1$. As was the case at the transition occurring at $U_c/t \sim 4.2$ for $V = 0$, we

expect this transition to coincide with a metal-insulator transition for molecular solids formed of C_{20} . More generally, for $U/t \leq 3$ and $V/t \leq 0.2$, we find that the spin, charge and pairing correlations fall off rapidly even in the presence of NN Coulomb repulsion. It is an interesting open question if molecular solids formed of C_{20} , in particular away from half-filling, would display non-trivial order for $V > V_c$. The answer to this question would be numerically demanding and we have therefore left it for future work.

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